

REVISED STRUCTURE OF PSEUDODISTOMIN A, A PIPERIDINE
ALKALOID ISOLATED FROM THE OKINAWAN TUNICATE
PSEUDODISTOMA KANOKO

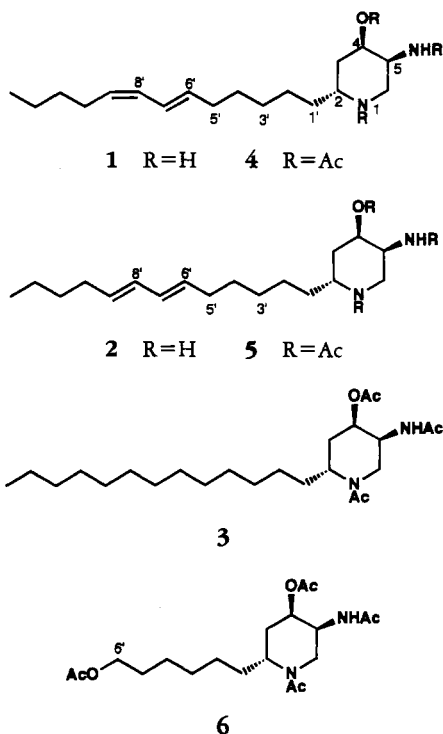
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ABSTRACT.—Pseudodistomin A [1], a piperidine alkaloid isolated from the Okinawan tunicate *Pseudodistoma kanoko*, was previously assigned with a 3'E,5'Z-diene in the side-chain; it is now shown to possess a 6'E,8'Z-diene. Ozonolysis of the triacetate of 1 afforded the same product as that obtained from the triacetate of the 6',8'-diene pseudodistomin B [2], indicating that the diene position of 1 is the same as that of 2.

During our studies on bioactive substances from marine organisms, we previously isolated two piperidine alkaloids, pseudodistomins A [1] and B [2], from extracts of the Okinawan tunicate *Pseudodistoma kanoko* Tokioka & Nishikawa (family Polyclinidae) (1). These compounds were the first piperidine alkaloids obtained from marine sources, and exhibited cytotoxic and calmodulin-antagonistic activity. The structures of pseudodistomins A and B were initially proposed as 1 and 2, respectively, but with a 3',5'-diene unit in the side-chain. The structure of the piperidine nucleus portion was unambiguously established by synthesis of tetrahydroacetyl derivatives [3] achieved by three research groups (racemate: 2,3; optically active form: 4,5), while the structure of the diene position of pseudodistomin B was later revised to 2, having a 6',8'-diene in the side-chain (6). As a result, the diene position of pseudodistomin A was also questioned, but remained undetermined because a natural specimen was unavailable. We recently reexamined the toluene-soluble fraction of extracts of this tunicate, and succeeded in reisolating pseudodistomin A as its acetate [4]. This report revises the structure of pseudodistomin A, which also proved to possess a 6',8'-diene in the side-chain on the basis of an ozonolysis experiment.

The tunicate *P. kanoko*, collected off the Ie Islands, Okinawa, Japan, was ex-



tracted with MeOH-toluene (3:1), and the extract was partitioned between toluene and H₂O.¹ The toluene-soluble material was subjected to Si gel flash cc (CHCl₃-*n*-BuOH-AcOH-H₂O, 1.5:6:1:1). A part of the fraction containing the alkaloid mixture was acetylated (Ac₂O/pyridine)

¹The H₂O-soluble fraction was further extracted with CHCl₃. In the previous report (1), pseudodistomins A and B were isolated from this CHCl₃-soluble fraction.

and the acetate mixture was purified by reversed-phase hplc (ODS, 85% MeOH) to give the acetate of pseudodistomin A [4] together with the acetate of pseudodistomin B [5] in 0.012 and 0.016% yield, respectively, based on wet wt of the tunicate.

To determine the position of its conjugated diene moiety, pseudodistomin A acetate was subjected to ozonolysis by the same procedures used for pseudodistomin B acetate [5] (6), namely, treatment with ozone followed by reduction with NaBH₄ and acetylation with Ac₂O and pyridine. The product was identical with the tetraacetate 6 previously obtained from pseudodistomin B acetate [5] (6), on the basis of tlc, ¹H-nmr, and eims analyses. From this result, the position of the diene moiety of pseudodistomin A was revised as occurring at the C-6',C-8' position [1]. In the HOHAHA nmr spectrum (mixing time: 67 msec) of pseudodistomin A acetate, the terminal methyl protons

(H₃-13') showed a clear correlation with the allylic methylene protons at δ_H 2.15 (H₂-10'), while no correlation was observed between H₃-13' and the other allylic methylene protons at δ_H 2.07 (H₂-5'). The latter methylene (C-5') was adjacent to an *E* olefin whereas the former methylene (C-10') was vicinal to a *Z* olefin, which was revealed by the ¹H-¹H COSY nmr spectrum and coupling constant data of 4 (cross-peaks: H₂-5'/H-6', H₂-5'/H-7', H₂-10'/H-8', and H₂-10'/H-9'; *J*_{6',7'} = 15 Hz and *J*_{8',9'} = 11 Hz). Pseudodistomin A [1] must therefore have a 6'*E*,8'*Z*-diene in the side-chain. In Table 1, the revised assignments of ¹H-nmr data of pseudodistomin A acetate [4] are described together with those of pseudodistomin B acetate [5].

EXPERIMENTAL

ANIMAL MATERIAL.—The tunicate *Pseudodistoma kanoko* was collected off the Ie Islands (at depths of 5 to 10 m), Okinawa, Japan, in 1985, and

TABLE 1. Revised ¹H-Nmr Assignments for Pseudodistomin A and B Acetates [4 and 5].

Position	Compound	
	4	5
H-2	4.93 br s	4.86 br s
H-2 ^a	4.00 br s	3.97 br s
H ₂ -3	1.75 m	1.74 m
H-4	5.15 m	5.11 m
H-5	4.31 br s	4.34 br s
H-5 ^a	4.51 br s	4.47 br s
H-6 _{ax}	3.27 br d (<i>J</i> =14.5 Hz)	3.26 br d (<i>J</i> =14.5 Hz)
H-6 _{ax} ^a	2.91 br d (<i>J</i> =14.5 Hz)	2.88 br d (<i>J</i> =14.5 Hz)
H-6 _{eq}	3.97 br d (<i>J</i> =14.5 Hz)	3.87 br d (<i>J</i> =14.5 Hz)
H-6 _{eq} ^a	4.61 br d (<i>J</i> =14.5 Hz)	4.58 br d (<i>J</i> =14.5 Hz)
H ₂ -1'	1.60 m	1.55 m
H ₂ -2'-H ₂ -4'	1.20-1.40 br s	1.20-1.40 br s
H ₂ -5'	2.07 m	2.03 m
H-6'	5.65 m	5.46-5.57 m
H-7'	6.29 dd (<i>J</i> =15 and 11 Hz)	5.91-5.99 m
H-8'	5.93 t (<i>J</i> =11 Hz)	5.91-5.99 m
H-9'	5.30 m	5.46-5.57 m
H ₂ -10'	2.15 m	2.03 m
H ₂ -11'-H ₂ -12'	1.20-1.40 br s	1.20-1.40 br s
H ₃ -13'	0.89 t (<i>J</i> =7 Hz)	0.86 t (<i>J</i> =7 Hz)
-NH-	5.60 br s	6.18 br d (<i>J</i> =8 Hz)
Ac	2.00, 2.04, and 2.06	2.00, 2.00, and 2.01

^aSignals of the minor conformer.

kept frozen until used. The voucher specimen (TN-84) was deposited at the Graduate School of Human Informatics, Nagoya University.

EXTRACTION AND ISOLATION.—The MeOH-toluene (3:1, 1 liter \times 2) extract of this tunicate (0.4 kg) was partitioned between toluene (500 ml \times 4 and 1 M NaCl (1 liter). The toluene-soluble fraction was evaporated under reduced pressure to give a crude residue (1.68 g), a portion of which (0.76 g) was subjected to Si gel cc (2.4 \times 36 cm) with CHCl₃-*n*-BuOH-AcOH-H₂O (1.5:6:1:1). The fraction (224 mg) eluting from 930 ml to 1100 ml was ninhydrin-positive and contained a mixture of piperidine alkaloids. A part (65 mg) of this fraction was treated with Ac₂O (0.6 ml) and pyridine (0.6 ml) at room temperature overnight, and the acetate mixture was purified by reversed-phase hplc (Develosil ODS-5, 5 μ m, 10 \times 250 mm; eluent, 85% MeOH; flow rate, 2.0 ml/min; detection, uv at 230 nm) to give pseudodistomin A acetate (**4**, 6.2 mg, *R*, 19.4 min) together with pseudodistomin B acetate (**5**, 8.6 mg, *R*, 21.1 min).

OZONOLYSIS OF PSEUDODISTOMIN A ACETATE [4].—A solution of pseudodistomin A acetate (**4**, 1.0 mg) in MeOH (0.5 ml) was bubbled with O₃ at -78° for 30 min. After the removal of excess ozone by bubbling N₂, a solution of NaBH₄ (10 mg) in MeOH (0.4 ml) was added and the mixture was stirred for 1 h at 0 $^\circ$. After addition of 1.5 ml of 1 M potassium phosphate buffer (pH 7.0), the reaction mixture was evaporated under reduced pressure. The residue was treated for Ac₂O (0.5 ml) and pyridine (1 ml) at room temperature for 12 h.

Brine (0.5 ml) was added and the mixture was extracted with EtOAc (0.5 ml \times 3). The organic phase was dried over MgSO₄ followed by evaporation under reduced pressure to give a residue (0.9 mg), which was subjected to tlc, ¹H-nmr, and eims analysis, and proved to be identical with the tetraacetate **6** (6) [tlc, Si gel, MeOH-CHCl₃ (1:9), visualized with anisaldehyde, *R*_f 0.55; ¹H nmr (CDCl₃, 270 MHz) (6); eims (70 eV) *m/z* 325 (M-CH₃CONH₂)⁺ (62), 282 (31), 181 (40), 139 (50) and 80 (100).

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